

1. (Currently Amended) A composition comprising ~~calcium phosphate platelets~~ which exhibits at least one of a monetite, or predominant monetite or deficient apatite structure and ~~which has~~ wherein the calcium phosphate platelets have a length of between 250 nm and 800 nm.
2. (Currently Amended) The composition comprising ~~calcium phosphate platelets~~ according to claim 1, ~~characterized in that the length is between 250 nm and 600 nm, preferably wherein the calcium phosphate platelets have a length of between 250 nm and 400 nm.~~
3. (Currently Amended) The composition comprising ~~calcium phosphate platelets~~ according to claim 1, ~~either of the preceding claims, characterized in that its~~ wherein the calcium phosphate platelets have a thickness is of between 1 nm and 40 nm.
4. (Currently Amended) The composition comprising ~~calcium phosphate platelets~~ according to ~~one of Claims 1 to~~ claim 3, wherein a plurality of the platelets have a monetite structure exhibiting a chemical shift of between -1.4 ppm and -1 ppm, ~~as measured by phosphorus-31 MAS NMR, assigned to the monetite structure.~~
5. (Currently Amended) The composition comprising ~~calcium phosphate platelets~~ according to ~~one of Claims 1 to~~ claim 3, wherein a plurality of the platelets have an apetite structure exhibiting a chemical shift of between 3 ppm and 3.4 ppm, measured by phosphorus-31 MAS NMR, ~~assigned to the apatite structure.~~
6. (Currently Amended) The composition comprising ~~calcium phosphate platelets~~ according to claim 1, ~~one of Claims 1 to 5, characterized in that it exhibits~~ wherein the calcium phosphate platelets have a calcium to phosphorus molar ratio of between 0.95 and 1.4, ~~for the monetite structure, preferably of between 1.1 and 1.3, and of between 0.95 and 1.4 for the monetite structure mixed with the brushite and apatite structure, preferably of between 1.1 and 1.3.~~
7. (Currently Amended) The composition comprising ~~calcium phosphate platelets~~ according to ~~one of Claims 1 to~~ claim 3, characterized in that it exhibits wherein the calcium phosphate platelets have a calcium to phosphorus molar ratio of between 1.25 and 1.67, ~~for the deficient apatite structure, preferably of between 1.3 and 1.6.~~

8. (Currently Amended) An aqueous dispersion comprising calcium phosphate platelets according to claim 3, one of Claims 1 to 7.

9. (Currently Amended) A colloidal dispersion obtained by suspending comprising calcium phosphate platelets according to one of Claims 1 to 7 claim 3 in an aqueous solution containing the presence of a dispersing agent.

10. (Currently Amended) Process A method for preparing the calcium phosphate platelets wherein the calcium phosphate platelets have a length of between 250 nm and 800 nm comprising the steps of: according to Claims 1 to 6, characterized in that it comprises the following stages:

- i) preparing a solution of calcium salt and adjusting the pH of the solution to a selected value of which is between 4 and 6;
- ii) adding a phosphate solution to the solution obtained in stage step i) over a period of time of between 30 minutes and 4 hours, so as to obtain a calcium to phosphorus molar ratio of between 1 and 2.5, wherein and while keeping the pH is maintained constant at a the selected value of between 4 and 6;
- iii) heat treating the dispersion solution obtained in stage step ii) at a temperature of between 50°C and 95°C;
- iv) separating the calcium phosphate platelets formed from the dispersion solution obtained in stage step iii);

and in that it uses, in wherein in at least one of stages steps i) or ii), the solutions further comprise comprising an ammonium ions.

11. (Currently Amended) Process A method for preparing the calcium phosphate platelets wherein the calcium phosphate platelets have a length of between 250 nm and 800 nm comprising the steps of: according to Claims 1 to 3 and 7, characterized in that it comprises the following stages;

- i) preparing a solution of calcium salts, and adjusting the pH of which is to a selected value of between 4 and 6;
- ii) adding a phosphate solution to the solution obtained in stage step i) over a period of time of between 30 minutes and 4 hours, so as to obtain a calcium to phosphorus molar ration of between 1 and 2.5, wherein and while keeping the pH is maintained constant at a the selected value of between 4 and 6;

- iii) heat treating the dispersion solution obtained in stage step ii) at a temperature of between 50°C and 95°C;
- iv) adjusting the pH of the dispersion solution obtained in stage step iii) to a value of between 8 and 9.5;
- v) separating the calcium phosphate platelets formed from the dispersion solution obtained in stage step iv);

and in that it uses, wherein in at least one of stages i) or ii), the solutions comprising an further comprise ammonium ions.

12. (Currently Amended) Process The method according to either of claims claim 10 and 11, characterized in that wherein the solution of calcium salts is a CaCl₂ or Ca(NO₃)₂ solution.

13. (Currently Amended) Process The method according to one of claim 10 to 12, characterized in that wherein the concentration of calcium salts in the solution is between 1M and 2.5M, preferably between 1.25M and 1.75M.

14. (Currently Amended) Process The method according to one of Claims claim 10 to 13, characterized in that wherein the phosphate salt solution is a solution of ammonium phosphate or of sodium phosphate, in particular of (NH₄)₂(HPO₄) or (NH₄) (H₂PO₄).

15. (Currently Amended) Process The method according to one of Claims claim 10 to 14, characterized in that wherein the calcium to phosphorous molar ratio is between 1.3 and 1.7.

16. (Cancelled)

17. (Currently Amended) Process The method according to Claims claim 10 to 16, characterized in that wherein the temperature of the heat treatment in stage step iii) is between 60°C and 90°C.

18. (Cancelled)

19. (New) The method according to claim 11, wherein the solution of calcium salts is a CaCl₂ or Ca(NO₃)₂ solution.

20. (New) The method according to claim 11, wherein the concentration of calcium salts in the solution is between 1M and 2.5M.

21. (New) The method according to claim 11, wherein the phosphate solution is a solution of (NH₄)₂(HPO₄) or (NH₄) (H₂PO₄).

22. (New) The method according to claim 11, wherein the calcium to phosphorous molar ratio is between 1.3 and 1.7.